

Chapter – Rate of Chemical Reactions and Chemical Equilibrium.

1. What is the unit of the rate of reaction?
2. If the value of PH is greater than 7, will it be acid or base?
3. What do you mean by instantaneous rate of Chemical reaction?
4. Define endothermic and exothermic reaction? Give one example each.
5. Define strong and weak electrolytes?
6. How can you explain term acid and base on the basis of hewis concept?
7. What is Photochemical reaction? Give one example.
8. Find the PH of a solution formed by mixing .03 mole hydrochloric acid and 3 litre of water?
9. Explain the factors which effects rate of reaction?

Or

Define chemical equilibrium with an example? How you can prove that chemical equilibrium is always dynamic in nature.

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